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(1986), Solubility and Related Properties, Marcel Dekker, Inc., N.Y. On page 5 the authors state, "The volume fraction of a solvent in a solution is defined as the volume of solvent in the solution divided by the total volume of solution." The authors further note that intermolecular volume effects and variations due to temperature are usually ignored. In the "Handbook of Pharmaceutical Excipients" abbreviations and units of measurement are discussed on p. xv. "v/v" is defined as volume in volume and "w/v" is defined as weight in volume. Wade, A. and Weller, P. Ed. (1994) Handbook of Pharmaceutical Excipients. This interpretation is further exemplified in the Maniatis cloning manual. For example, 10% ammonium persulfate is described as 1 gm of ammonium persulfate in a total volume of 10 mls. Ten percent SDS is described as 100 gms of SDS in a total of 1 liter. See Sambrook et al. (1989) Molecular Cloning: A laboratory manual, Cold Spring Harbor Laboratory Press, Appendix B.10-B.12.

In view of the remarks and amendments provided herein as well as the remarks provided in Applicants first response to the Examiner's final rejection, it is respectfully submitted that the rejections and objections have been overcome. The claim amendments presented above do not add new matter. Entry of the amendment, reconsideration and withdrawal of the rejection are therefore requested.

Respectfully submitted,

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